# Inactivation of Individual Ca<sup>2+</sup>-Binding Sites in the Paired EF-Hand Sites of Parvalbumin Reveals Asymmetrical Metal-Binding Properties<sup>†</sup>

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ABSTRACT: Previously a rat parvalbumin mutant protein  $PV_{F102W}$  was constructed with a reporter Trp at position 102 in the middle of the hydrophobic center [Pauls, T. L., et al. (1993) J. Biol. Chem. 268, 20897-20903]. In the present study three new parvalbumin mutant proteins, derived from PV<sub>F102W</sub> and containing alterations at positions essential for Ca2+ binding in either one of the two Ca2+-binding sites  $(PV_{-CD} \text{ and } PV_{-EF})$  or in both  $(PV_{-CD/-EF})$ , were expressed and purified. With the flow dialysis method it was established that both  $PV_{-CD}$  and  $PV_{-EF}$  bind 1  $Ca^{2+}$  with affinity constants  $K_{Ca}$  of  $1.1 \times 10^7$  and 3.2× 106 M<sup>-1</sup>, respectively. Mg<sup>2+</sup> binding, monitored by equilibrium gel filtration in the absence of Ca<sup>2+</sup>, showed that both mutants bind 1 Mg<sup>2+</sup> with  $K_{\text{Mg}} = 8 \cdot 10^4$  for PV<sub>-CD</sub> and 3 × 10<sup>3</sup> M<sup>-1</sup> for PV<sub>-EF</sub>. Compared to the parameters of the parent mutant PV<sub>F102W</sub> (two sites with equal affinities of  $2.7 \times 10^7$  and  $3 \times 10^4$ M-1 for Ca2+ and Mg2+, respectively), these data indicate that inactivation of the EF site, much more than of the CD site, impairs divalent cation binding. The binding of Ca<sup>2+</sup> and Mg<sup>2+</sup> is mutually exclusive, indicative of a  $Ca^{2+}/Mg^{2+}$  mixed site. However, as for  $PV_{F102W}$ , the  $K_{Mg}$  values obtained from the competition equation are approximately 40-fold lower than the affinities measured by direct binding. PV\_CD/-EF binds neither Ca<sup>2+</sup> nor Mg<sup>2+</sup>. Trp fluorimetry revealed that in the three mutant PVs the residue Trp-102 is deeply buried in the hydrophobic core. The fluorescence spectra of the metal-free, Ca<sup>2+</sup>, and Mg<sup>2+</sup> forms of PV<sub>CD</sub> are very similar to those of the parent mutant PV<sub>F102W</sub>. In contrast, the spectra of PV<sub>-EF</sub> indicate the presence of a much less organized hydrophobic core, especially in the metal-free form. The Trp fluorescence of PV<sub>CD/-EF</sub> is not sensitive to Ca<sup>2+</sup> or Mg<sup>2+</sup> and resembles that of metal-free PV<sub>-EF</sub>. The comparison of ligand-induced UV difference spectra of PV\_CD and PV\_EF confirms that Trp-102 is located in a hydrophobic core and that inactivation of the EF loop, but not of the CD loop, strongly destabilizes the protein in the absence of Ca<sup>2+</sup>, thus explaining the impaired cation affinity of PV<sub>-EF</sub>. Our data indicate that in rat parvalbumin the EF-hand site is the principal structural nucleus of the paired domain.

High-affinity Ca<sup>2+</sup> binding to many intracellular proteins occurs at a well organized center called the paired EF-hand domain, which consists of two helix-loop-helix motifs assembled onto a two-site domain with a two-fold symmetry axis. In each EF-hand two amphiphilic helices, oriented at approximately 90°, flank a 12-residue loop, which provides six Ca<sup>2+</sup>-coordinating oxygen ligands, defined X, Y, Z, -Y, -X, and -Z from the N- to the C-terminus (Kretsinger, 1987). The loop also contains a short  $\beta$ -pleated sheet, which is important for the pairing of two EF-hands (Shaw et al., 1990; Kay et al., 1991). Usually both sites of the paired EF-hand domain are active, with a distance of 11 Å between the two bound Ca<sup>2+</sup> ions, but this is by no means obligatory: cardiac troponin C (Van Eerd & Takahashi, 1975) and yeast calmodulin (Matsuura et al., 1993) possess a low-affinity site paired to an abortive site; calbindin-D28K (Gross et al., 1993), the regulatory light chain of myosin (Bagshaw & Kendrick-Jones, 1979), and all invertebrate sarcoplasmic Ca<sup>2+</sup>-binding proteins [for a review, see Cox (1990)] display a high-affinity site paired to an abortive one. The paired configuration has been demonstrated directly in the three-dimensional structure of the sarcoplasmic Ca<sup>2+</sup>-binding proteins of *Nereis* (Cook *et al.*, 1991) and of amphioxus (Cook *et al.*, 1993) and the regulatory light chain of myosin (Rayment *et al.*, 1993).

In spite of the inherent symmetry of the paired Ca2+-binding domain, the two sites do not necessarily have the same Ca<sup>2+</sup>binding characteristics. The paired domain of oncomodulin contains one high-affinity Ca<sup>2+</sup>/Mg<sup>2+</sup> mixed site and one Ca<sup>2+</sup>specific site (Hapak et al., 1989; Cox et al., 1990). In contrast. the closely homologous parvalbumin (50% sequence identity) has two paired sites, both of the Ca<sup>2+</sup>/Mg<sup>2+</sup> mixed type. The structural requirements for the difference in affinity and selectivity of the different types of domains are not known, but small, punctual differences in the sequences of the Ca<sup>2+</sup>binding domains influence the metal-binding properties (MacManus et al., 1989; Trevino et al., 1991). There are also indications that the sites within a paired domain influence each other and that this interference is not symmetrical. Studies on mutant forms of  $\alpha$ -actinin (Witke et al., 1993) or of calmodulin (Maune et al., 1992) show that the C-terminal site within each paired Ca<sup>2+</sup>-binding domain is more important in maintaining the tertiary structure of this domain than the N-terminal site, whereas in troponin C it appears to be the opposite (Brito et al., 1993). This raises the question of whether conformational information is directional, i.e., whether it propagates from one defined site to the other, and not vice versa. Nothing is known about the hinges and levers in this

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<sup>&</sup>lt;sup>1</sup> The term paired EF-hand domain or two-site domain is used to underline the notion that the individual Ca<sup>2+</sup>-binding sites do not form a stable structural element, but as a pair they form a thermodynamically stable, independent structural unit (Shaw et al., 1990).

intramolecular information transduction process.

To get more insight into the interactions between two homologous EF-hands in the paired Ca<sup>2+</sup>-binding domain, we used rat parvalbumin (PV)<sup>2</sup> with its one functional pair of Ca<sup>2+</sup>/Mg<sup>2+</sup> mixed sites as a model. In a previous study (Pauls et al., 1993) Phe-102 was replaced by a unique Trp in the central hydrophobic core in rat PV, leading to a mutant protein, PV<sub>F102W</sub>, with a suitable optical probe inside the protein. Its cation-binding properties were very similar to those of the recombinant unmodified PV<sub>WT</sub> and of rat wildtype parvalbumin isolated from muscle. In the present study we further modified this mutant by point mutations in order to change essential features of PV, such as its ability to bind two Ca2+ ions. The Ca2+- and Mg2+-binding characteristics and ligand-induced conformational changes in two mutant parvalbumins with a single functional Ca2+-binding site are described in this report.

#### MATERIALS AND METHODS

Mutagenesis, Expression, and Purification. Site-directed mutagenesis (Kunkel et al., 1987) was carried out in order to modify the rat PV cDNA in the vector pGEMEX (Promega) which had been used previously for cloning and expression of recombinant rat PV (Pauls & Berchtold, 1993). The DNA template pGEMEX-PV<sub>F102W</sub> used for mutagenesis contained an initial amino acid replacement Phe-102-Trp for monitoring structural changes upon metal binding (Pauls et al., 1993). On top of this mutation either one or both of the functional Ca<sup>2+</sup>-binding sites (CD or EF domain) were inactivated. Two synthetic oligonucleotides, CD<sub>51/62</sub> and EF<sub>90/101</sub>, containing mismatches in codon 51 (GAC  $\rightarrow$  GCC) and codon 62 (GAG  $\rightarrow$  GTG) of the C loop or in codon 90 (GAC  $\rightarrow$  GCC) and codon 101 (GAA  $\rightarrow$  GTG) of the EF loop, respectively, were synthesized: CD<sub>51/62</sub>, 5'-CAC-ATT-CTG-GCC-AAA-GAC-AAA-AGT-GGC-TTC-ATT-GAG-GAG-GAT-GTG-CTG-GGG-TCC-3'; EF<sub>90/101</sub>, 5'-GCT-GCT-GGA-GCC-AAG-GAC-GGG-GAC-GGC-AAG-ATT-GGG-GTT-GAA-GTG-TGG-TCC-ACT-3'. Modified codons in the oligonucleotides are underlined. These oligonucleotides were used for replacing the first and last amino acids, Asp and Glu, in the Ca<sup>2+</sup>-binding loops by the two nonpolar amino acids Ala and Val, respectively. The resulting PV mutant proteins PV<sub>D51A,E62V,F102W</sub> and PV<sub>D90A,E101V,F102W</sub> are named PV<sub>CD</sub> and PV-EF. In addition a third PV mutant protein, PV<sub>D51A,E62V,D90A,E101V,F102W</sub>, in which both loops were altered, was produced by mutating PV\_CD with the oligonucleotide EF<sub>90/101</sub> for obtaining the double-defective mutant PV<sub>-CD/-EF</sub>. The mutant proteins were identified by the dideoxy chain termination sequencing method (Sanger et al., 1977).

Expression and purification of all mutant proteins were carried out according to the procedure previously described (Pauls et al., 1993), with some modifications. In the last purification step PV\_CD and PV\_EF were purified to homogeneity by DEAE-Sephacel ion-exchange chromatography (Pharmacia) with 20 mM imidazole and 2 mM EDTA, pH 7.8, as equilibration buffer. PV\_CD and PV\_EF eluted as pure proteins during subsequent washing with equilibration buffer. PV\_CD/-EF eluted in the flowthrough of this column, which was then dialyzed against 10 mM Tris-HCl, pH 8.5, and 1 mM EDTA and loaded onto a QAE-Sephadex A-25 ion-

exchange column (Pharmacia).  $PV_{-CD/-EF}$  was eluted in a 0-50 mM NaCl gradient. Pure protein-containing fractions were identified by SDS-PAGE and silver staining. For subsequent studies the pure fractions were dialyzed in  $H_2O$  and either lyophilized or stored at -80 °C.

Protein and Metal Ion Determination and Metal Removal. For removal of contaminating metals and for complete equilibration of the protein in the assay buffer, the proteins were precipitated with 3% trichloroacetic acid and then passed through a 40 × 1 cm Sephadex G-25 column equilibrated in 50 mM Tris-HCl, pH 7.5, and 150 mM KCl (buffer A). The assay buffer was freed of contaminating metals by passage over a column of EDTA-agarose (Haner et al., 1984). Total Ca<sup>2+</sup> and Mg<sup>2+</sup> concentrations were determined with a Perkin-Elmer Cetus Instruments 2380 atomic absorption spectrophotometer as previously described (Pauls et al., 1993). The protein concentration was determined from the ultraviolet absorption spectrum using molar extinction coefficients at 280 nm of 6400, 6500, and 6900 M<sup>-1</sup> cm<sup>-1</sup> for PV<sub>-CD</sub>, PV<sub>-EF</sub>, and PV<sub>-CD</sub>/-EF, respectively.

Cation Binding. Ca<sup>2+</sup> binding was measured at 25 °C by the flow dialysis method of Colowick and Womack (1969) in buffer A containing 0, 1, 5, and 10 mM MgCl<sub>2</sub>. Protein concentrations were 10-30  $\mu$ M. Treatment of the raw data was as described by Cox et al. (1990). From this and previous studies we estimate that the accuracy of the binding constants obtained from flow dialysis experiments is around 80%. Mg<sup>2+</sup> binding to parvalbumin in the absence of Ca<sup>2+</sup> was measured by the equilibrium gel filtration method of Hummel and Dryer (1962). A Sephadex G-25 column (0.7  $\times$  50 cm) was equilibrated in buffer A supplemented with 50 mM EGTA and the indicated concentrations of MgCl2. One-half to 1 mL of 50-200 μM metal-free parvalbumin was applied to the column. In the eluant Mg2+ concentrations were determined by atomic absorption, and protein concentrations, by ultraviolet absorption.

The intrinsic metal-binding constants were determined with a Scatchard plot. The  $Mg^{2+}$  effect on the  $Ca^{2+}$ -binding curves was evaluated with the competition equation  $K_{Ca}/K_{Ca.app} = 1 + K_{Mg}[Mg^{2+}]$ , where  $K_{Ca}$  and  $K_{Ca.app}$  are the  $Ca^{2+}$ -binding constants in the absence and presence of  $Mg^{2+}$ , respectively, and  $K_{Mg}$  is the  $Mg^{2+}$ -binding constant in the absence of  $Ca^{2+}$ .

Spectrofluorimetry. Emission fluorescence spectra were taken with a Perkin-Elmer LS-5B spectrofluorimeter interfaced with a computer. The measurements were carried out on metal-free solutions of 7  $\mu$ M mutant PVs in buffer A at room temperature with excitation wavelength at 280 nm and slits of 2.5 nm. Two millimolar MgCl<sub>2</sub> and 1 mM CaCl<sub>2</sub> were added subsequently to obtain the Mg2+ and Ca2+ forms, respectively. Alternatively, guanidine hydrochloride up to 4 M was added to obtain the fluorescence characteristics of the fully denatured form. All of the spectra were normalized to an optical density of 0.05 at 280 nm. Mg<sup>2+</sup> titrations of the mutant proteins following the fluorescence change were carried out under experimental conditions identical with those described above. Spectra were recorded after each increment, and the signal change was normalized between 0 (no Mg2+ added) and 1 (20 mM Mg2+ added).

Conformational Changes Monitored by Near-Ultraviolet Absorption. UV-absorption spectra and difference spectra were measured with a Perkin-Elmer Cetus Instruments Lambda 5 spectrophotometer at room temperature. The metal-free protein was equilibrated in buffer A. Difference spectra were taken on solutions with an optical density at 280 nm of 1, corrected for dilutions and normalized to molar

 $<sup>^2</sup>$  Abbreviations: PV, parvalbumin; PV\_CD, PV with the substitutions D51A/E62V/F102W (abortive CD site); PV\_EF, PV with the substitutions D90A/E101V/F102W (abortive EF site); PV\_CD/-EF, PV with the substitutions D51A/E62V/D90A/E101V/F102W (both sites abortive).

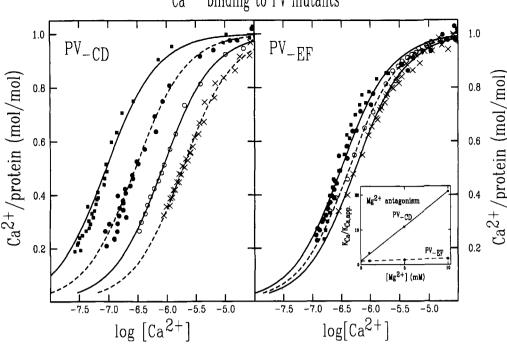


FIGURE 1: Ca<sup>2+</sup> binding to recombinant PV<sub>CD</sub> and PV<sub>EF</sub>. Ca<sup>2+</sup> binding was measured by the flow dialysis method at 25 °C. The PV concentrations were  $20-30 \mu M$ . Mg<sup>2+</sup> concentrations were 0 ( $\blacksquare$ ), 1 ( $\bullet$ ), 5 (O), or 10 mM (×). The curves were calculated with the following  $K_{ass}$  constants:  $1.1 \times 10^7$ ,  $3.3 \times 10^6$ ,  $1.0 \times 10^6$ , and  $5.2 \times 10^5$  M<sup>-1</sup> for PV<sub>CD</sub>;  $3.2 \times 10^6$ ,  $2.3 \times 10^6$  and  $1.9 \times 10^6$  M<sup>-1</sup> for PV<sub>EF</sub>. Inset: Ratio of the intrinsic Ca<sup>2+</sup>-binding constant in the absence of Mg<sup>2+</sup> over the apparent Ca<sup>2+</sup>-binding constant in the presence of Mg<sup>2+</sup> as a function of [Mg<sup>2+</sup>] for PV<sub>CD</sub> ( $\blacksquare$ ) and PV<sub>EF</sub> ( $\bullet$ ).

concentrations. For spectra involving the metal-free form a correction was carried out to take account of the contamination by Ca<sup>2+</sup>, assuming a linear dependence of the signal on Ca<sup>2+</sup>. Ca<sup>2+</sup> and Mg<sup>2+</sup> titrations were done in buffer A (in Mg<sup>2+</sup> titrations 20  $\mu$ M EGTA was present to complex contaminating Ca<sup>2+</sup>). The titrations with Ca<sup>2+</sup> are stoichiometric, and therefore the difference signals  $D_{290-298nm}$  and  $D_{283-288nm}$  were plotted as a function of added Ca<sup>2+</sup> per mole of protein; due to the much lower affinity of the proteins for Mg<sup>2+</sup>, the Mg<sup>2+</sup> titrations are nonstoichiometric and the signal changes were plotted as a function of the concentration of added Mg<sup>2+</sup>.

#### **RESULTS**

Mutagenesis, Expression, and Purification of Recombinant Proteins. Three different mutant proteins were isolated. Each of the variants contained the initial Phe-102-Trp exchange and additional substitutions in their Ca2+-binding loops, as described under Materials and Methods. Isopropyl Dthiogalactopyranoside induced expression of mutant PVs in Escherichia coli strain JM109 (DE3) resulted in protein production similar to that previously described for PV<sub>WT</sub> and PV<sub>F102W</sub> (Pauls et al., 1993) with 25-30% mutant protein per total bacterial protein for PV<sub>CD</sub> and 20% for PV<sub>EF</sub> and PV\_CD/-EF. Nevertheless, since bacteria expressing PV\_EF and PV\_CD/-EF grew more slowly, the total amounts of mutant protein per liter of bacterial culture were approximately 2-fold smaller for  $PV_{-EF}$  and  $PV_{-CD/-EF}$  than for  $PV_{F102W}$  and  $PV_{-CD}$ . PV\_EF and PV\_CD/-EF showed a higher mobility in SDS-PAGE when compared to PV<sub>F102W</sub> (data not shown).

The first steps of purification for all mutant proteins were done as previously described (Pauls et al., 1993) with DEAE-Sephacel ion exchange as the last step. However, the removal of two negative charges in PV\_CD and PV\_EF resulted in a weaker binding to the ion exchanger. The mutant proteins

eluted after approximately 3 column volumes of subsequent washing with 20 mM imidazole and 2 mM EDTA, pH 7.8. The purity of the cleanest fractions was >95% with final yields of 45 and 8 mg/(liter of bacterial culture) for PV\_CD and PV\_EF, respectively. PV\_CD/-EF, lacking four negative charges compared to PV\_F102w, did not bind at all to DEAE-Sephacel. This mutant protein had to be further purified using the stronger ion exchanger QAE-Sephadex A-25 equilibrated in 10 mM Tris-HCl and 1 mM EDTA, pH 8.5. PV\_CD/-EF was purified to >95% homogeneity with a final yield of 6 mg/ (liter of bacterial culture).

Direct Binding Studies. Figure 1 shows the Ca2+-binding isotherms of PV<sub>-CD</sub> and PV<sub>-EF</sub> in the presence of 0, 1, 5, and 10 mM Mg<sup>2+</sup> as measured by the flow dialysis method. As predicted from the sequence, the mutant proteins display only one site for Ca2+, without indications of an additional lowaffinity site (conclusion confirmed by difference spectrophotometry). In the absence of Mg<sup>2+</sup> the affinity constants for  $Ca^{2+}$  are 1.1 × 10<sup>7</sup> and 3.2 × 10<sup>6</sup> M<sup>-1</sup> for PV<sub>-CD</sub> and PV<sub>-EF</sub>, respectively. Since the parent mutant PV<sub>F102W</sub> has two equivalent sites with  $K_{Ca}$  of  $2.7 \times 10^7$  M<sup>-1</sup> (Pauls et al., 1993), it must be concluded that in the new mutant proteins the pairing of an active site to an abortive one diminishes the Ca2+ affinity of the active site to a variable degree, depending on which site was destroyed. This asymmetry is also manifested in the antagonistic effect of  $Mg^{2+}$  on  $Ca^{2+}$  binding to the residual site in  $PV_{-CD}$  and  $PV_{-EF}$ .  $Mg^{2+}$  affects the Ca<sup>2+</sup> affinities of the two mutant proteins in very different ways. For PV<sub>CD</sub> the apparent constant for  $Ca^{2+}$ ,  $K_{Ca,app}$ , decreases to  $5 \times 10^5 \,\mathrm{M}^{-1}$  in 10 mM Mg<sup>2+</sup>. Interpreted with the competition equation (see Materials and Methods), the affinity constant for Mg<sup>2+</sup> amounts to  $2 \times 10^3$  M<sup>-1</sup> (inset of Figure 1), which is very similar to the calculated competition constant in PV<sub>F102W</sub>. In contrast, the affinity of the mutant PV<sub>-EF</sub> for Ca<sup>2+</sup> is weakly influenced by Mg<sup>2+</sup> (Figure 1).

## Mg<sup>2+</sup> binding to PV mutants

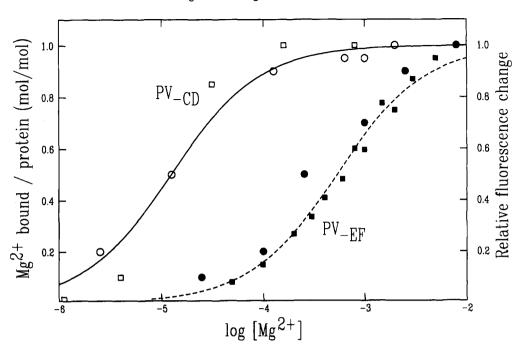


FIGURE 2:  $Mg^{2+}$  binding to recombinant  $PV_{-CD}$  (open symbols) and  $PV_{-EF}$  (closed symbols) measured by the Hummel-Dryer method (O,  $\bullet$ ) and by the Trp fluorescence change ( $\Box$ ,  $\blacksquare$ ) at room temperature. The following estimates of  $K_{Mg}$  were obtained: Hummel-Dryer method, 8  $\times$  10<sup>4</sup>  $M^{-1}$  for  $PV_{-CD}$  and 3  $\times$  10<sup>3</sup>  $M^{-1}$  for  $PV_{-EF}$ ; fluorescence titration, 2  $\times$  10<sup>3</sup>  $M^{-1}$  for  $PV_{-EF}$ . The curves were calculated with the  $K_{ass}$  constants 8  $\times$  10<sup>4</sup> ( $\longrightarrow$ ) and 2  $\times$  10<sup>3</sup>  $M^{-1}$  (- - -), respectively.

With the competition equation one obtains a  $K_{\text{Mg}}$  value of 75  $M^{-1}$ . This value is even lower than the affinity constant of calmodulin for  $Mg^{2+}$  (Milos *et al.*, 1986).

Direct  $Mg^{2+}$ -binding studies, carried out with the Hummel–Dryer method, are shown in Figure 2. Both mutant proteins bind 1  $Mg^{2+}$ , and the values of the affinity constants amount to  $8 \times 10^4~M^{-1}$  for  $PV_{-CD}$  and  $3 \times 10^3~M^{-1}$  for  $PV_{-EF}$ . The magnitudes of these constants have been confirmed by fluorimetric and difference spectrophotometric titrations (see below). The 40-fold higher affinity for  $Mg^{2+}$  determined here as compared to the affinity calculated from the competition behavior has also been observed in wild-type parvalbumin and in  $PV_{F102W}$  and cannot yet be explained.

The Ca<sup>2+</sup>- and Mg<sup>2+</sup>-binding capacity of the mutant PV<sub>-CD/-EF</sub> was assayed with the Hummel-Dryer method: in the presence of 100  $\mu$ M free Ca<sup>2+</sup> and 1 mM free Mg<sup>2+</sup> no noticeable amount of metal ion was bound to the mutant protein (data not shown).

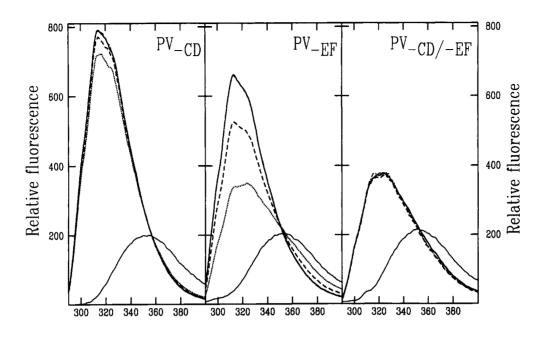
Fluorescence Characteristics. Trp emission fluorescence spectra of the different forms of PV<sub>-CD</sub>, PV<sub>-EF</sub>, and PV<sub>-CD/-EF</sub> are shown in Figure 3. As previously described for mutant PV<sub>F102</sub>w, the spectra of the new mutant proteins show strongly blue-shifted maxima and display a high quantum yield, indicating that in these mutant proteins, too, the Trp residue is deeply buried in the hydrophobic core. This Trp residue is completely unmasked in the presence of guanidine hydrochloride (thin solid line in Figure 3). With their maximum at 313 nm, the spectra of the Ca2+, Mg2+, and metal-free forms of PV\_CD do not differ much from each other and are very similar in shape and intensity to the corresponding spectra of PV<sub>F102W</sub>. Those of the PV<sub>-EF</sub> mutant protein, however, are very different: the metal-free form has a maximum at 325 nm. Upon binding of  $Mg^{2+}$ , the maximum moves to 313 nm but the intensity is still 1.5-fold lower than that of  $Mg^{2+}$ loaded PV\_CD. Only upon binding of Ca2+ did the fluorescence

intensity of  $PV_{-EF}$  approach that of  $PV_{F102W}$  and  $PV_{-CD}$ . The Trp fluorescence spectrum of  $PV_{-CD/-EF}$  is  $Ca^{2+}$ -independent; its shape and intensity suggest that in this inactive mutant protein the environment of the Trp is the same as in metal-free  $PV_{-EF}$ .

The  $Mg^{2+}$  titration of  $PV_{-EF}$  followed by the fluorescence change of Trp (Figure 2) gives an affinity constant,  $K_{Mg}$ , of  $2 \times 10^3$  M<sup>-1</sup>, which resembles the value from direct binding studies, i.e.,  $3 \times 10^3$  M<sup>-1</sup>. The Trp fluorescence change for  $PV_{-CD}$  is very small (Figure 3), but it is obvious that it occurs in the  $10-100~\mu$ M range ( $\Box$  in Figure 2), as expected from the direct binding study.

Difference Spectrophotometry. Figure 4 shows the difference spectra of PV\_CD and PV\_EF induced by Ca2+ and Mg<sup>2+</sup> and, for the purpose of the analyses, the difference spectra between the metal-free or Ca2+ form and the guanidine hydrochloride denatured protein. Qualitatively the spectra resemble those obtained from PV<sub>F102W</sub>, suggesting a similar overall geometry of the hydrophobic core. Importantly, the signal change of the transition from the unfolded to the metalfree conformation is much greater for PV<sub>CD</sub> than for PV<sub>EF</sub>, indicating that metal-free PV-EF is much closer to the denatured state than metal-free PV-CD. Conversely, the difference spectra for the transitions from the metal-free to the Ca2+- and Mg2+-loaded forms are smaller for PV\_CD than for PV\_EF. Thus PV\_CD has the characteristics of the mutant PV<sub>F102W</sub>, whereas PV<sub>-EF</sub> is much less structured in the metalfree form and recovers the PV<sub>F102W</sub> structure only upon binding of Ca<sup>2+</sup>. It is worth noticing that the calculated total signal change from the denatured to the Ca2+ form (thin line in Figure 4) is quite similar for the two mutant proteins, thus confirming the fluorescence experiments which indicate that the Ca2+-saturated conformations of both mutant proteins are quite similar. The Mg2+ conformations are intermediate between those of the metal-free and the Ca2+ form. For

### Trp fluorescence spectra of PV mutants



emission wavelength (nm)

FIGURE 3: Fluorescence spectra of PV<sub>CD</sub>, PV<sub>EF</sub>, and PV<sub>CD/EF</sub>. The protein concentration was  $7 \mu M$ .  $Ca^{2+}$ ,  $Mg^{2+}$ , or guanidine hydrochloride was added up to 2.5 mM, 5 mM, or 3 M, respectively. The spectra were normalized as described in Materials and Methods. Thick solid line,  $Ca^{2+}$  form; dashed line,  $Mg^{2+}$  form; dotted line, metal-free form; dashed and dotted line,  $Ca^{2+} + Mg^{2+}$  form; thin solid line, guanidine hydrochloride denatured form. Note that the dashed and dotted line nearly coincides with the thick solid line.

### UV difference spectra of PV mutants

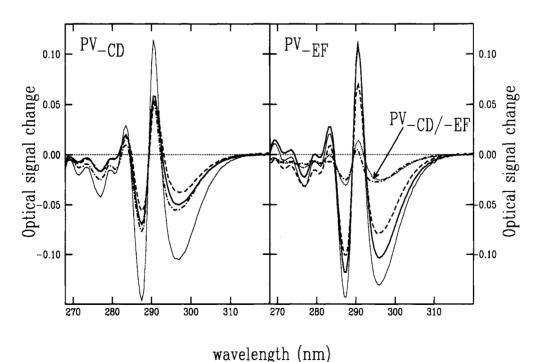


FIGURE 4:  $Ca^{2+}$ - or  $Mg^{2+}$ -induced difference spectra of the recombinant PV mutant proteins  $PV_{-CD}$  and  $PV_{-EF}$  at room temperature. The protein concentrations were  $150~\mu M$ . To the sample cell was added  $Ca^{2+}$ ,  $Mg^{2+}$ , or guanidine hydrochloride up to 2.5 mM, 5 mM, or 3 M, respectively, whereas in the reference cell an identical dilution was performed with buffer A. The spectra were normalized as described in Materials and Methods. Thick solid line,  $Ca^{2+}$  form versus metal-free form; dashed line,  $Mg^{2+}$  form versus metal-free form; dashed and dotted line, metal-free form versus denatured form; thin solid line,  $Ca^{2+}$  form versus denatured form (calculated). The spectrum of the metal-free form versus the denatured form of the mutant  $PV_{-CD/-EF}$  is also shown in the right panel as a thin dotted line (arrow).

# titration of PV mutants

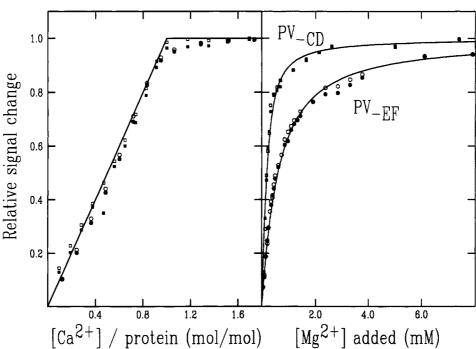


FIGURE 5: Ca<sup>2+</sup> dependence (A) or Mg<sup>2+</sup> dependence (B) of the difference signals  $D_{290-298nm}$  (open symbols) and  $D_{283-288nm}$  (filled symbols) of the mutants PV<sub>CD</sub> ( $\square$ ,  $\blacksquare$ ) and PV<sub>EF</sub> ( $\bigcirc$ ,  $\bullet$ ). The measurements were carried out as detailed in Figure 4. In the left panel the optic cell was 0.5 cm wide and the protein concentrations were 391 and 304 µM for PV\_CD and PV\_EF, respectively. In the right panel the optic cell was 1 cm wide, the protein concentrations were 168 and 181 µM for PV<sub>-CD</sub> and PV<sub>-EF</sub>, respectively, and the solid lines were calculated with K<sub>Mg</sub> values of  $2 \times 10^4$  and  $2 \times 10^3$  M<sup>-1</sup> for PV<sub>-CD</sub> and PV<sub>-EF</sub>, respectively.

PV<sub>-CD/-EF</sub> the UV difference spectral changes from the unfolded to the folded form indicate that this mutant protein resembles closely PV\_EF in its less structured, metal-free conformation (Figure 4, dotted line), which is in agreement with the fluorimetric data.

The Ca<sup>2+</sup> titrations as followed by the relative absorption difference signal (Figure 5A) show that mutants PV<sub>CD</sub> and PV<sub>-EF</sub> bind 1 mol of Ca<sup>2+</sup> per mole of protein with a high affinity. The Mg<sup>2+</sup> dependency of the absorption difference (Figure 5B) allows one to calculate the  $K_{Mg}$  values, using the following equation:

$$\Delta S/\Delta S_{\text{max}} = C_{\text{PV}} K_{\text{Mg}} [\text{Mg}^{2+}]/(1 + K_{\text{Mg}} [\text{Mg}^{2+}])$$

where  $\Delta S$  and  $\Delta S_{\text{max}}$  are the measured and maximal signal changes and  $C_{PV}$  is the parvalbumin concentration. The  $K_{Mg}$ value for PV<sub>-EF</sub>, i.e.,  $2 \times 10^3$  M<sup>-1</sup>, is similar to that obtained in Figure 2. The value for PV<sub>-CD</sub>,  $2 \times 10^4$  M<sup>-1</sup>, is rather low when compared to those obtained by direct binding and fluorescence titration, but given the high affinity and protein concentration used, estimation of the upper limit of the  $K_{Mg}$ value is not precise.

### DISCUSSION

In the present study mutant parvalbumins derived from a previous mutant, PV<sub>F102W</sub>, and containing either one (PV<sub>-CD</sub> and PV<sub>-EF</sub>) or two (PV<sub>-CD/-EF</sub>) abortive Ca<sup>2+</sup>-binding sites in their C-terminal tandem domain have been analyzed. The highly conserved positions 1 (Asp) and 12 (Glu) in the 12residue Ca<sup>2+</sup>-binding loop, which provide essential oxygen ligands for Ca<sup>2+</sup>, were replaced by the nonpolar residues Ala and Val, respectively. NMR showed that in the loops of pike PV these residues exhibit the biggest changes upon metal binding (Blancuzzi et al., 1993). The same replacements in all three functional loops in yeast calmodulin resulted in weak Ca<sup>2+</sup> binding when position 12 was changed, whereas no Ca<sup>2+</sup> binding was detectable when both positions 1 and 12 were substituted (Geiser et al., 1991). Replacements of position 1 alone,  $D \rightarrow A$ , also prevented the binding of  $Ca^{2+}$  to yeast mutant calmodulins (Geiser et al., 1991), whereas the same replacement for single sites in troponin C was not enough to inactivate the mutated site (Negele et al., 1992). Therefore, we have chosen to substitute both positions in each Ca<sup>2+</sup>binding loop. As expected, PV<sub>-CD</sub> and PV<sub>-EF</sub> bind only 1 Ca<sup>2+</sup> and 1 Mg<sup>2+</sup>, whereas PV<sub>-CD/-EF</sub> binds no metal ion at

PV<sub>F102W</sub>, PV<sub>-CD</sub>, PV<sub>-EF</sub>, and PV<sub>-CD/-EF</sub> all contain a single substitution, Phe-102-Trp, with the fluorescent indole group positioned in the center of the protein equally distant from both Ca<sup>2+</sup>-binding sites. All mutant PVs display the very strong blue-shifted Trp fluorescence spectra with unusually high quantum yield. The difference spectra clearly show that in the new mutant proteins, as in PV<sub>F102W</sub>, the Trp is hidden in the hydrophobic core; but a positively charged electronwithdrawal group, present in the hydrophobic core, affects the electron density of the indole ring of Trp102 (Ilich et al., 1988). The comparison of the fluorescence and UV difference spectra of PV\_EF with those of PV\_CD leads to the conclusion that in PV\_EF the hydrophobic core surrounding the Trp residue is much less structured than in PV\_CD.

One of the most intriguing observations in our study is that, despite the structural symmetry in the two-site domain of parvalbumin, these two sites do not have the same structural importance within the paired EF-hand domain. In the wildtype protein and in PV<sub>F102W</sub> the two sites appear to be equivalent with the same affinity for Ca2+ and the same antagonistic effect of Mg<sup>2+</sup> on Ca<sup>2+</sup> binding (Pauls et al., 1993). But important differences show up when either the N-terminal or the C-terminal site is modified so that it cannot bind Ca<sup>2+</sup> anymore. When the CD site is inactivated, the remaining EF site still has a high affinity for Ca2+ (2.5-fold lower than in parent PV<sub>F102W</sub>) and for Mg<sup>2+</sup> (2-fold higher than in PV<sub>F102W</sub>). Moreover, this mutant protein displays a conformation and metal-induced conformational changes which are very similar to those in PV<sub>F102W</sub> (Pauls et al., 1993). In contrast, when the EF site is inactivated, the remaining CD site has a 10-fold lower affinity for Ca2+ and for Mg2+, the antagonistic effect of Mg2+ is strongly decreased, and conformational studies suggest that the hydrophobic core is much more accessible to polar solvent. Apparently, in the paired domain the EF site is structurally much more important than the CD site and constitutes the stable structural nucleus of the paired domain. When this structure is impaired, the CD site binds cations with lower affinity because the metalfree state is much more disorganized than that of the native

The mutant PV\_EF is also much less sensitive to Mg<sup>2+</sup> than the mutant PV<sub>-CD</sub>, which suggests that when the CD loop is not backed by a functional helix-loop-helix motif, its -Z ligand, Glu-62, is unable to assume the g-  $f\chi 1$  rotamer (Blancuzzi et al., 1993), or at least this rotamer is energetically much less favored. Glu-101 in the EF loop can do so, even in the mutant PV<sub>CD</sub>. The reason could be that the PV<sub>EF</sub> is already much less structured, and going to the highest level of structure is more difficult than for the mutant PV<sub>-CD</sub>.

An interesting comparison can be made between the structural properties of the mutant parvalbumins with inactive Ca<sup>2+</sup>-binding sites and those of other EF-hand-containing proteins with similar mutations. Studies on mutant calmodulins with single Ca2+-binding sites in any two of the paired domains (Maune et al., 1992a,b; Gao et al., 1993) show that mutations in sites II and IV prove to be more deleterious for Ca2+ binding than mutations in sites I and III. UV difference spectrophotometry of a unique Tyr-138 in site IV of the C-terminal domain indicates that site III mutants can partially reconstitute their hydrophobic environment, whereas this is effectively abolished in site IV mutants. In nonmuscle type  $\alpha$ -actinin, too, the inactivation of either one of the paired Ca<sup>2+</sup>-binding sites revealed the asymmetry in Ca<sup>2+</sup> binding: the site I mutant protein still binds Ca<sup>2+</sup> with high affinity, whereas the site II mutant shows a 500-fold lower affinity for Ca2+ than the wild-type protein (Witke et al., 1993). In the C-terminal half of cardiac troponin C the site III and site IV mutant proteins bind one  $Ca^{2+}$  with a  $K_{Ca}$ around 106 M<sup>-1</sup> (Negele et al., 1992), and this is sufficient to promote formation of the short  $\beta$ -sheet between the two sites and large changes in the environment of the aromatic residues (Brito et al., 1993), suggesting symmetry. But Ca<sup>2+</sup> binding at site III appears more important for the stability of the C-terminal domain (contrary to the case of parvalbumin; this study). Thermodynamic studies on calmodulin and skeletal muscle troponin C have previously shown that site III is the least stable Ca<sup>2+</sup>-binding site, having an ordered structure only in the presence of bound Ca<sup>2+</sup> (Tsalkova & Privalov, 1985). In conclusion, even when the affinities for Ca<sup>2+</sup> of the EF-hand sites in a symmetrical paired two-site domain are the same, inactivation studies systematically reveal a structurally dominant site. In parvalbumin, calmodulin, and  $\alpha$ -actinin this dominant site is at the C-terminal end; in troponin C it is at the N-terminal end.

The functional importance of parvalbumin as a buffer of Ca2+ and Mg2+ can now be evaluated by transfection of these mutant PVs in living cells. Table 1 shows the dissociation

Table 1: Dissociation Constants of PV\_CD, PV\_EF, PVF102W, and PVwT for Ca2+ and Mg2+ a

	PV <sub>-CD</sub>	PV <sub>-EF</sub>	PV <sub>F102W</sub> b	PV <sub>WT</sub> <sup>b</sup>
Ca2+ binding, µM				
in 0 Mg <sup>2+</sup>	0.1	0.3	0.04	0.04
in 1 mM Mg <sup>2+</sup>	0.3	0.3	0.2	0.2
Mg <sup>2+</sup> binding, mM				
from competition	0.5	13	0.5	0.5
from direct methods	0.012	0.3	0.025	0.03

<sup>&</sup>lt;sup>a</sup> Round numbers were given for all dissociation constants. <sup>b</sup> Taken from Pauls et al. (1993).

constants of the wild-type and mutant proteins for Ca2+ and Mg<sup>2+</sup> under physiological concentrations of Mg<sup>2+</sup> (taken as 1 mM), revealing that the four proteins display the same sensitivity to the Ca<sup>2+</sup> signal. PV<sub>-CD</sub>, PV<sub>F102W</sub> and PV<sub>WT</sub>, are supposed to be fully saturated with Mg<sup>2+</sup> at rest, whereas PV<sub>-EF</sub> would be mostly in a metal-free state. Therefore, upon an increase of intracellular Ca<sup>2+</sup> levels the former three proteins would first have to release their bound Mg<sup>2+</sup> in order to take up Ca<sup>2+</sup>, whereas PV<sub>-EF</sub> could immediately bind Ca<sup>2+</sup>. This allows the effects of Ca<sup>2+</sup>/Mg<sup>2+</sup> exchange and of Ca<sup>2+</sup> buffering to be studied separately. The mutant PV\_CD/-EF can yield indications of whether the role of parvalbumin is restricted to the above two actions or includes also divalent cation independent functions. First results with a transfected epidermal cell line show a decrease in mitotic rate, changes in morphology from epithelioid to fusiform, and an increase in motility after expression of PV<sub>WT</sub> (C. Andressen, personal communication). In contrast, no changes were found in cells expressing the Ca<sup>2+</sup>-binding-deficient mutant PV<sub>CD/-EF</sub> when compared to untransfected cells.

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### **REFERENCES**

Bagshaw, C. R., & Kendrick-Jones, J. (1979) J. Mol. Biol. 130, 317-336.

Bailey, J. E., Beaven, G. H., Chignell, D. A., & Gratzer, W. B. (1968) Eur. J. Biochem. 7, 5-14.

Blancuzzi, Y., Padilla, A., Parello, J., & Cavé, A. (1993) Biochemistry 32, 1302-1309.

Brito, R. M. M., Krudy, G. A., Negele, J. C., Putkey, J. A., & Rosevear, P. R. (1993) J. Biol. Chem. 268, 20966-20973.

Colowick, S. P., & Womack, F. C. (1969) J. Biol. Chem. 244,

Cook, W. J., Ealick, S. E., Babu, Y. S., Cox, J. A., & Vijay-Kumar, S. (1991) J. Biol. Chem. 266, 652-656.

Cook, W. J., Jeffrey, L. C., Cox, J. A., & Vijay-Kumar, S. (1993) J. Mol. Biol. 229, 461-471.

Cox, J. A. (1990) in Stimulus-response coupling: the role of intracellular calcium (Dedman, J. R., & Smith, V. L., Eds.) pp 83-1070, CRC Press, Boca Raton, FL.

Cox, J. A., Milos, M., & MacManus, J. P. (1990) J. Biol. Chem. *265*, 6633–6637.

Declerq, J. P., Tinant, B., Parello, J., & Rambaud, J. (1991) J. Mol. Biol. 220, 1017-1039.

Gao, Z. H., Krebs, J., VanBerkum, F. A., Tang, W.-J., Maune, J. F., Means, A. R., Stull, J. T., & Beckingham, K. (1993) J. Biol. Chem. 268, 20096-20104.

Geiser, J. R., van Tuinen, D., Brockerhoff, S. E., Neff, M. M., & Davis, T. N. (1991) Cell 65, 949-959.

- Gross, M. D., Gosnell, M., Tsarbopoulos, A., & Hunziker, W. (1993) J. Biol. Chem. 268, 20917-20922.
- Haner, M., Henzl, M. T., Raissouni, B., & Birnbaum, E. R. (1984) Anal. Biochem. 138, 229-234.
- Hapak, R. C., Lammers, P. J., Palmisano, W. A., Birnbaum, E. R., & Henzl, M. T. (1989) J. Biol. Chem. 264, 18751-18760.
- Hummel, J. P., & Dryer, W. J. (1962) Biochim. Biophys. Acta 63, 530-532.
- Ilich, P., Axelsen, P. H., & Prendergast, F. G. (1988) Biophys. Chem. 29, 341-349.
- Kay, L. E., Forman-Kay, J. D., McCubbin, W. D., & Kay, C. M. (1991) Biochemistry 30, 4323-4333.
- Kretsinger, R. H. (1987) Cold Spring Harbor Symp. Quant. Biol. 52, 499-510.
- Kunkel, T. A., Roberts, J. D., & Zakour, R. A. (1987) Methods Enzymol. 154, 367-382.
- Luan-Rilliet, Y., Milos, M., & Cox, J. A. (1992) Eur. J. Biochem. 208, 133-138.
- MacManus, J. P., Hutnik, C. M. L., Sykes, B. D., Szabo, A. G., & Williams, T. C. (1989) J. Biol. Chem. 264, 3470-3477.
- Matsuura, I., Kimura, E., Tai, K., & Yazawa, M. (1993) J. Biol. Chem. 268, 13267-13273.
- Maune, J. F., Klee, C. B., & Beckingham, K. (1992a) J. Biol. Chem. 267, 5286-5295.
- Maune, J. F., Beckingham, K., Martin, S. R., & Bayley, P. M. (1992b) *Biochemistry 31*, 7779-7786.
- Milos, M., Schaer, J.-J., Comte, M., & Cox, J. A. (1986) Biochemistry 25, 6279-6287.

- Negele, J. C., Dotson, D. G., Liu, W., Sweeney, H. L., & Putkey, J. A. (1992) J. Biol. Chem. 267, 825-831.
- Pauls, T. L., & Berchthold, M. W. (1993) Methods Enzymol. 217, 102-122.
- Pauls, T. L., Durussel, I., Cox, J. A., Clark, I. D., Szabo, A. G.,
  Gagné, S. M., Sykes, B. D., & Berchthold, M. W. (1993) J.
  Biol. Chem. 268, 20897-20903.
- Rayment, Y., Rypniewski, W. R., Schmidt-Bäse, K., Smith, R., Tomchik, D. R., Benning, M. M., Winkelmann, D. A., Wesenberg, G., & Holden, H. M. (1993) Science 261, 50-58.
- Sanger, F. G., Nicklen, S., & Coulson, A. R. (1977) Proc. Natl. Acad. Sci. U.S.A. 74, 5463-5467.
- Shaw, G., Hodges, R. S., & Sykes, B. D. (1990) Science 249, 280-283.
- Trevino, C. L., Boschi, J. M., & Henzl, M. T. (1991) J. Biol. Chem. 266, 11301-11308.
- Tsalkova T. N., & Privalov, P. L. (1985) J. Mol. Biol 181, 533-544.
- Van Eerd, J. P., & Takahashi, K. (1976) Biochemistry 15, 1171-1180.
- Williams, T. C., Corson, D. C., Oikawa, K., McCubbin, W. D., Kay, C. M., & Sykes, B. D. (1986) *Biochemistry* 25, 1835– 1846.
- Witke, W., Hofmann, A., Köppel, B., Schleicher, M., & Noegel, A. A. (1993) J. Cell Biol. 121, 599-606.
- Wnuk, W., Cox, J. A., & Stein, E. A. (1981) in Calcium and Cell Function (Cheung, W. Y., Ed.) Vol. 2, pp 243–278, Academic Press, New York.